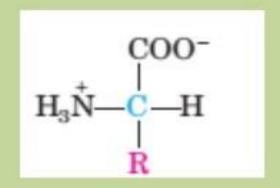
## AMINO ACIDS BY:NEHA MEGHANI DEPARTMENT OF BIOCHEMISTRY

#### **Amino Acid**

- It's a group of organic compounds containing two functional groups – amino (-NH2) and carboxyl group (-COOH)
- Its also called Zwitter Ion– both acidic and basic functional group (dipolar ion)



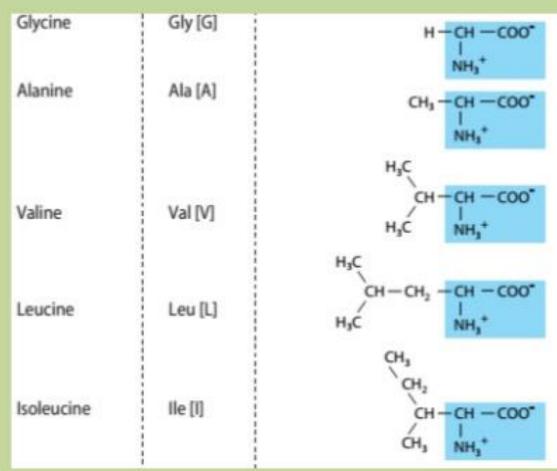
- This property is known as amphoteric and are often called ampholytes
- Neither humans nor any other higher animals can synthesize 10 of the 20 common amino acids – Essential Amino acids

#### Classification

- Amino acid has been classified under various ways
  - Structure
    - With side chain containing Aliphatic Side Chains
    - With Side Chains Containing Hydroxylic (OH) Groups
    - With Side Chains Containing Sulfur Atoms
    - With Side Chains Containing Acidic Groups or Their Amides
    - With Side Chains Containing Basic Groups
    - Containing Aromatic Rings
    - Imino Acid
  - Polarity
    - Non Polar
    - Polar
  - Nutritional
    - Essential and Non-essential

#### Side chain containing Aliphatic Side Chains

- Simplest amino acids
- Contains branched chain of hydrocarbons



#### Side Chains Containing Hydroxylic (OH) Groups

Serine	Ser [S]	CH <sub>2</sub> - CH - COO"     OH NH <sub>3</sub> +
Threonine	Thr [T]	CH <sub>3</sub> — CH — CH — COO"     OH NH <sub>3</sub> +
Tyrosine	Tyr [Y]	See below.

#### **Side Chains Containing Sulfur Atoms**



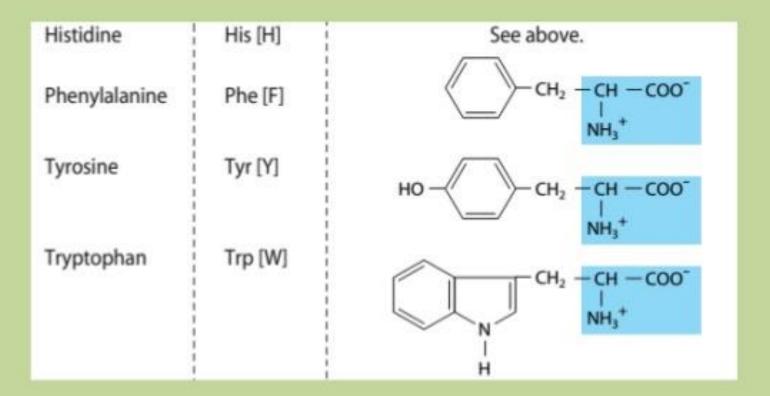
#### Side Chains Containing Acidic Groups or Their Amides

Aspartic acid	Asp [D]	<sup>-</sup> OOC — CH <sub>2</sub> — CH — COO <sup>-</sup>   NH <sub>3</sub> <sup>+</sup>
Asparagine	Asn [N]	$\begin{array}{c} H_2 N - C - C H_2 - C H - C O O^- \\ \parallel \\ O & N H_3^+ \end{array}$
Glutamic acid	Glu [E]	$OOC - CH_2 - CH_2 - CH - COO $
Glutamine	Gln [Q]	$\begin{array}{c} H_2 N - C - C H_2 - C H_2 - C H - C O O^{-1} \\ H_2 O & H_3^+ \end{array}$

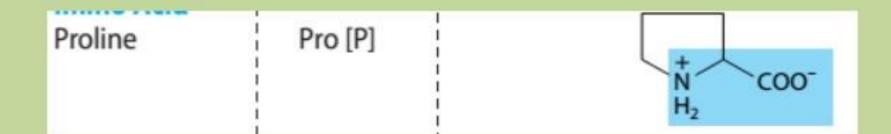
#### **Side Chains Containing Basic Groups**

Arginine	Arg [R]	$\begin{array}{c} H - N - CH_2 - C$
Lysine	Lys [K]	
Histidine	His [H]	$ \begin{array}{c c} & -CH - COO^{-} \\ HN & N \\ HN & NH_{3}^{+} \end{array} $

#### **Containing Aromatic Rings**



#### **Imino Acid**



#### **Classification : Polarity**

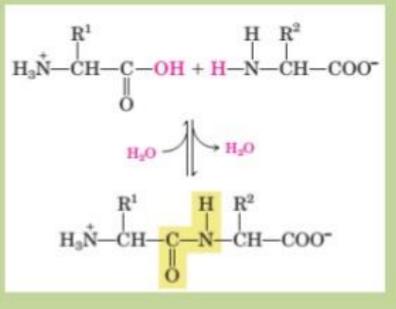
- Non-polar group : No charge on R group. Ex: Alanine, leucine. Isoleucine, valine, methionine, phenylalanine, tryptophan and proline
- Polar group
  - No charge on R : no charge on R but posses group such as hydroxyl, sulfhydryl and amide. Ex: Glycine, serine, threonine, cysteine, glutamine, asparigine and tyrsoine
  - Positive R- Lysine, arginine, and histidine
  - Negative R asparatic acid and glutamic acid

# **Essential Amino Acid (EAA)**

- It cant be synthesized in the body and therefore need to be supplied through diet
- Proper growth and maintenance of the individual
- Ex. Arginine, Valine, Histidine, Isoleucine, leucine, lysine, Methionine, Phenylalanine, Threonine, Tryphtophan
- Mnemonics : AV hill, MP TT
- Semi-essential amino acid: Adults can synthesize 2 amino acid and not by growing children. Ex: Arginine and histidine
- So in all 8 are essential and 2 semi essential

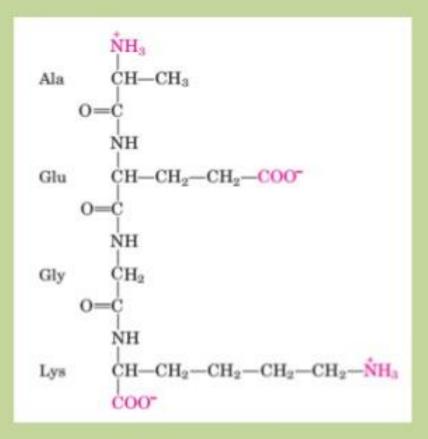
## PEPTIDES

- Two AA covalently joined through a substituted amide linkage – peptide bond
- Dehydration removal of H<sub>2</sub>O
  - OH<sup>-</sup> Carboxyl group of one AA
  - H<sup>+</sup> from amino group of another AA
- Example of a condensation reaction – common biological reactions



#### POLYPEPTIDES

- Two AA reacts to form dipeptides, Three AA can be joined by two peptide bonds to form a tripeptide and so on.
- Oligopeptide: When a few AA are joined by various peptide linkage
- When many amino acids are joined, the product is called a polypeptide.
- Proteins may have thousands of amino acid residues



#### Tetrapeptide

# THANK YOU